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The lab manual contains background information and procedures for the experiments you will perform as part of your General Chemistry II course " CHE1402. Along with concepts and chemistry covered in the lecture, the laboratory portion of the course will present some additional chemistry, both theoretical and practical (e.g. water analysis).

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Adding Na₂SO₄. 4 Color change after adding Na₂SO₄ solution: ____Based on the color change, the reaction shifted to the (circle one): left right. Explain the direction of the shift using Le Châtelier's principle: Adding NaHSO₄. 4 Color change after ... General Chemistry II Lab Manual

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General Chemistry II: Lab Manual is written for a second semester General Chemistry lab course. The experiments are ordered to follow the presentation that most instructors use at Middlesex County College. General Chemistry II: Lab Manual:-utilizes Microsoft Excel® for data manipulation with a spreadsheet.

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General Chemistry, Laboratory Manual CHM 1410L KOMAR UNIVERSITY OF SCIENCE AND TECHNOLOGY (KUST) 2 Preface and Acknowledgments The first version of General Chemistry I Lab Manual was drafted by Ms. Khalida Jaffar and Dr. Khalid Omer. It was used in the academic year, 2012-2013.

[General Chemistry CHM 1410L](#)

GENERAL CHEMISTRY LABORATORY RULES 1. You are responsible for all safety rules in the laboratory. 2. You have to wear a LAB COAT and safety goggles and also have a lab manual and calculator while performing experiments in the laboratory. Whitout them you are not

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Solution for General Chemistry II Laboratory Manual, 2019 Revision 133 Determination of Some Thermodynamic Data for the Dissolution of Borax Pre-l ab 1) Why is!

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2HI(g) " H₂(g) + I₂(g)Eq (1)If mainly reactant is present at equilibrium, the system is said to lie close to the reactants' side of the reaction; likewise the system may lie close to the products' side. The position of the equilibrium is characterized by an equilibrium constant, K_c, where the subscript "[c]" is for concentration.

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